

MIT OpenCourseWare
<http://ocw.mit.edu>

1.061 / 1.61 Transport Processes in the Environment
Fall 2008

For information about citing these materials or our Terms of Use, visit: <http://ocw.mit.edu/terms>.

Problem 1

Two airplanes are conducting tests to estimate the coefficient of diffusion in a stagnant atmosphere. The first airplane, flying 1 km above the earth's surface, instantaneously releases 1 kg of trace gas. The second airplane flies through the cloud 60 minutes later and measures a maximum concentration of 0.03 mg/L within the cloud.

Estimate the coefficient of diffusion within this region of the atmosphere.

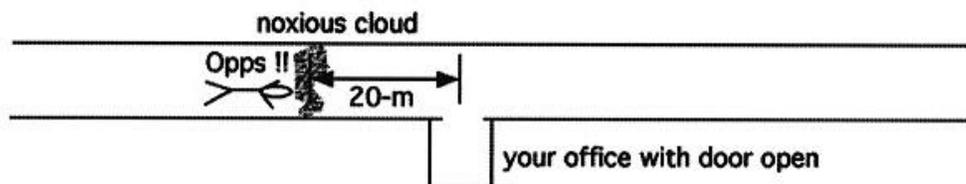
Hint 1 Can you assume uniform concentration in any direction?

Hint 2 What assumption can you make about the air currents?

Hint 3 What assumption must you make about the coefficient of diffusion?

Problem 2

A friend of yours is coming to meet you at your office, which is located mid-way along a very long (100 m) hallway. In preparation for a small prank, he is carrying a vial containing 10-g of a noxious smelling gas. He trips 20-m before reaching your office door, and the vial breaks. The gas rapidly mixes vertically and horizontally within the hallway, which is 2-m wide and 3-m high. The human nose will detect the gas at concentrations greater than $10 - \mu\text{g}/\text{l}$. Assume an isotropic diffusion, $D = 0.05 \text{ m}^2/\text{s}$.



- What governing equation describes the evolution of gas concentration in the hall?
- At what time after the spill do you smell the gas?
- When does the smell, as perceived by humans, disappear from the hallway?

Problem 3

A drop of red and a drop of blue dye, each 1 mg, are released 10 cm apart into a layer of stagnant fluid between two plates. The plates are $1 \text{ m} \times 1 \text{ m}$ in area and are 5 mm apart. The dye drops are released at the center of the plate area and mix rapidly across the fluid layer, i.e. between the plates. The molecular diffusion of the dyes are: $D_{\text{red}} = 10^{-5} \text{ cm}^2 \text{ s}^{-1}$ and $D_{\text{blue}} = 4 \times 10^{-5} \text{ cm}^2 \text{ s}^{-1}$.

The human eye can detect the color of the dye at concentrations of 10 g/L. No reactions occur between the two dyes, but at locations where the two dyes coexist and are both above the visible threshold, the mixture will appear purple.

- (a) While both clouds are fully visible ($C > 10 \text{ g/L}$), which cloud appears larger, and by how much?

Hint 1 Make a sketch that defines the diameter of each cloud

- (b) At what time and at what location will the two dye clouds first appear to touch?

Hint 2 Simplify the governing equations with assumptions

Hint 3 Write the mathematical criterion for the condition when clouds first touch

- (c) At what time will the line connecting the release points be completely purple?

Hint 4 Define a mathematical criterion for this to occur