

2.76/2.760 Multiscale Systems Design & Manufacturing

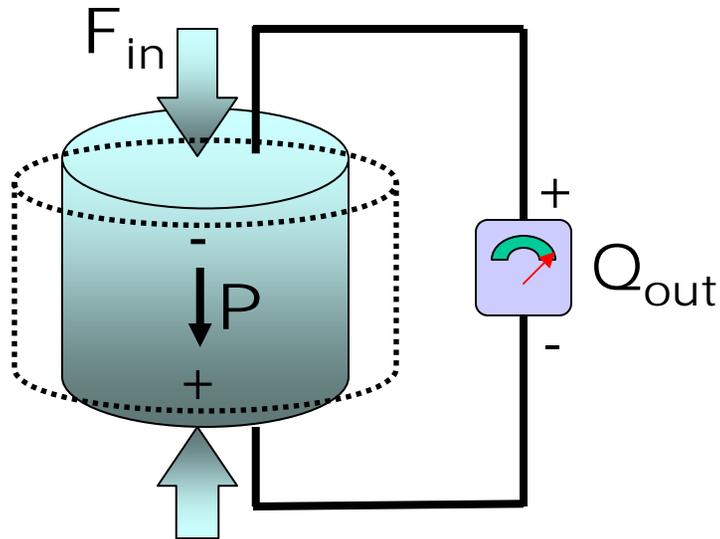
Fall 2004

Piezoelectric Transducers

What is Piezoelectricity ?

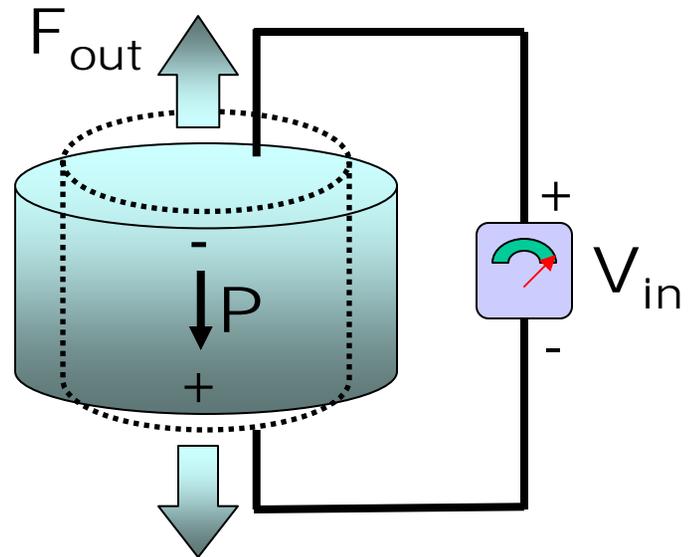
Jacques & Pierre Curie (1880)

**Electrical
Energy**



Lippmann (1881)

**Mechanical
Energy**



Piezoelectricity

- **Piezoelectric Ceramics**
- **Thin film fabrication**
- **Piezoelectric Transducers**

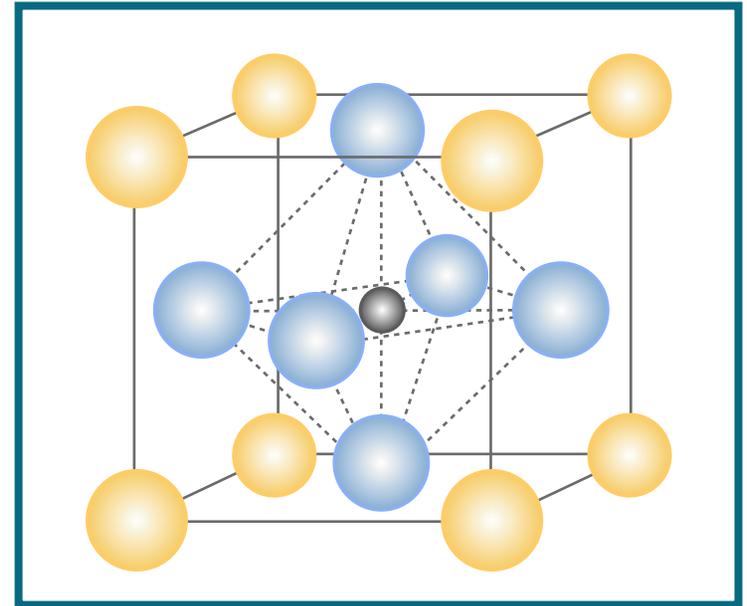


Figure by MIT OCW

Adapted from G. Binnig & H. Rohrer, Nobel Prize, 1986

Photos removed for copyright reasons.
Medical ultrasound, disk drive heads, naval submarine.

Piezoelectric Transducers

1-layer (unimorph)

2-layer

Diagrams removed for copyright reasons.

Source: Piezo Systems, Inc. "Introduction to Piezo Transducers."

<http://www.piezo.com/benedu.html>

Transducers

multi-layer (multimorph)

Diagrams removed for copyright reasons.

Source: Piezo Systems, Inc. "Introduction to Piezo Transducers."

<http://www.piezo.com/bendedu.html>

Component schematic removed for copyright reasons.

Physik Instrumente Z-positioner P-882.10, in http://www.pi-usa.us/pdf/2004_PICatLowRes_www.pdf, page 1-45.

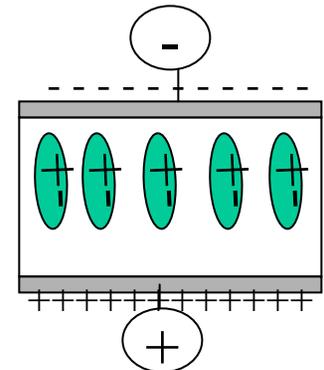
Dielectric Constant

- Dielectric materials, atoms are ionized, + and – charged, causing electric dipoles. – electric polarization
- Polarization, quantitatively,
electric dipoles/unit volume, C/m²
- Dielectric Displacement, D,
stored electric charge/unit area

$$\mathbf{D} = \epsilon_0 \mathbf{E} + \mathbf{P} = \epsilon_0 \mathbf{E} + K\epsilon_0 \mathbf{E}$$

ϵ_0 : Vacuum permittivity (8.854x10⁻¹² F/m)

K : Relative Permittivity or dielectric constant



Origins of Polarization

- Electronic Polarization

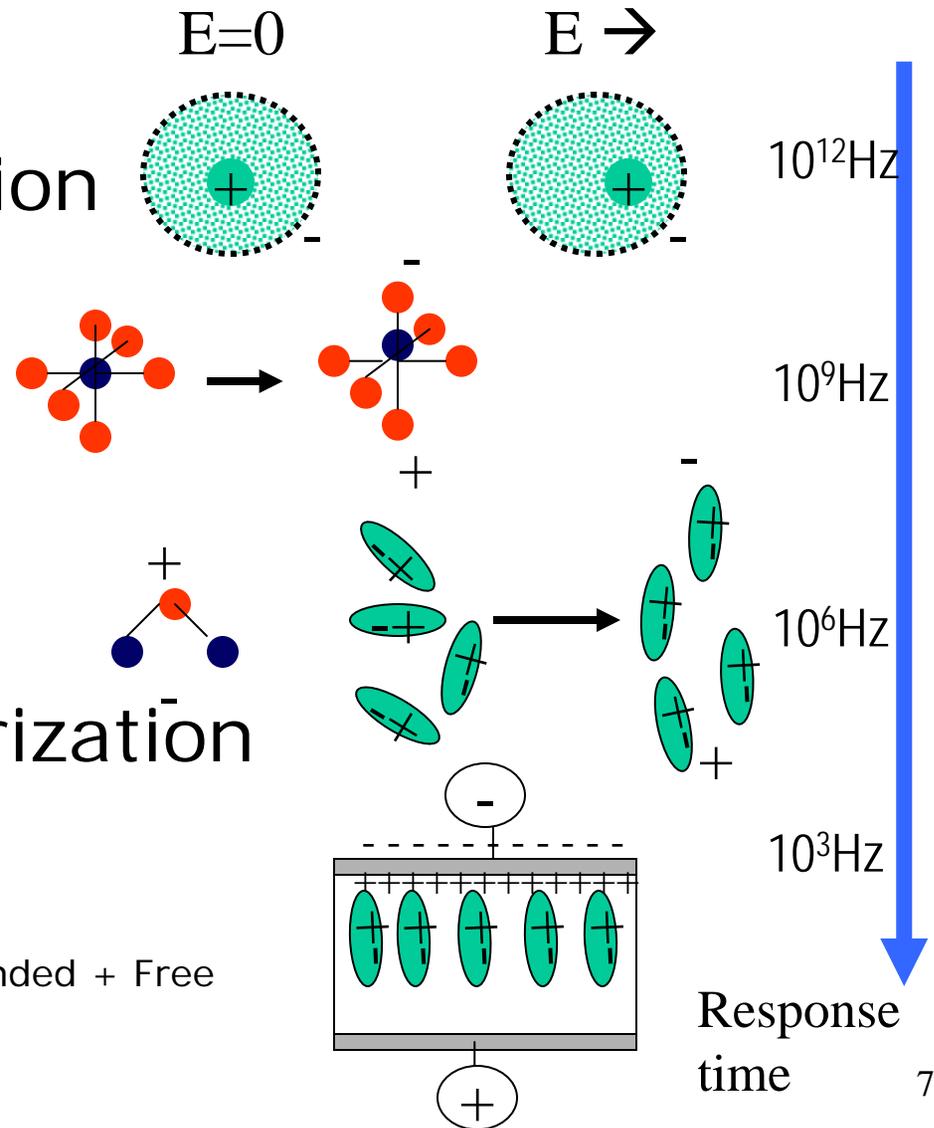
- Ionic Polarization

– Ionic crystals, NaCl..

- Dipolar Polarization

– HCl

- Space Charge Polarization

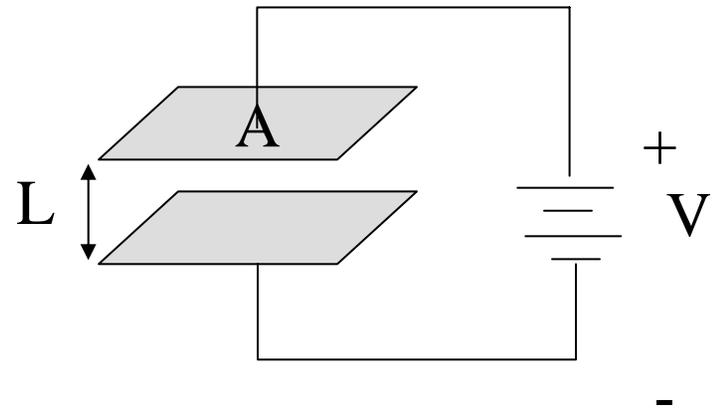


Capacitance

$$C = \frac{q}{V}$$

Capacitance of a parallel plate capacitor

$$C = \epsilon\epsilon_0 \frac{A}{L}$$



Relative Dielectric Constant (K), or permittivity

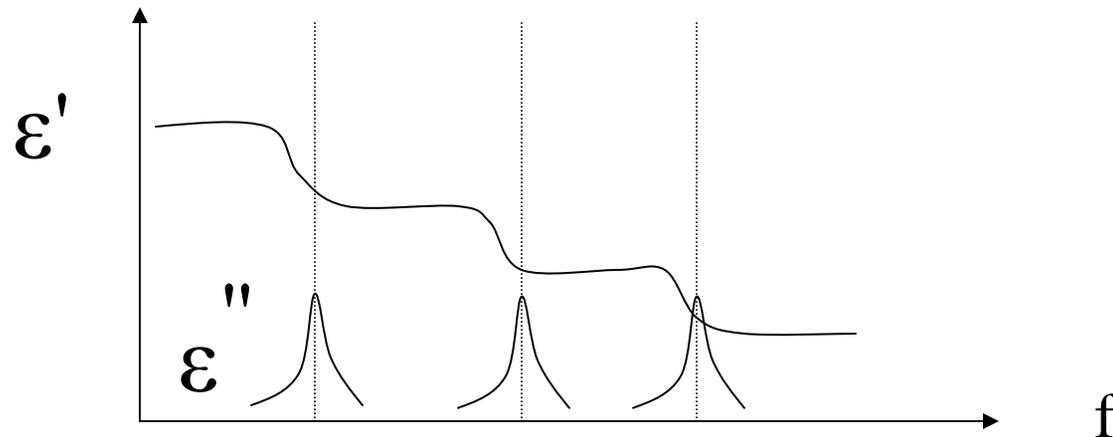
$$\epsilon = \frac{C}{C_{Vac}}$$

Dielectric Constants

- Water 81.1
- Silicon 11.8
- Glass 6.9
- Epoxy 4.0
- Ice 3.0
- Polyethylene 2.3
- Air 1.000576
- BaTiO₃ 6000
- KNbO₃ 700
- Rochelle salt 170
- PZT 1200

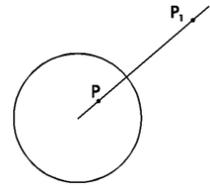
Dielectric Loss

- Frequency dependency of polarization
- When τ and f are in the same order, energy absorbed and phase lag occurs.
- Dielectric Loss, $\tan\delta = \epsilon'' / \epsilon'$
- Dielectric breakdown: 60kV/cm for PZT



Ferroelectricity

- Ionic and electronic polarization
- S^2P or d^0 cations
- Crystal structure with no inversion center
- Ferroelectricity;
 - Spontaneous polarization wrt external electric field
 - Reversible, Remanent polarization, hysteresis
- Pyroelectricity; strong variation in polarization with temperature
- Piezoelectricity; large mechanical strain response to applied electrical field



Perovskite Phase

- PZT, $(\text{Pb}(\text{Zr}_{1-x}\text{Ti}_x)\text{O}_3)$, the most important piezoelectric material.

$\text{PbZrO}_3:\text{PbTiO}_3 \rightarrow 52:48$

rhombohedral \rightarrow cubic \rightarrow tetrahedral

- Curie Temperature
- Pyrochlore phase

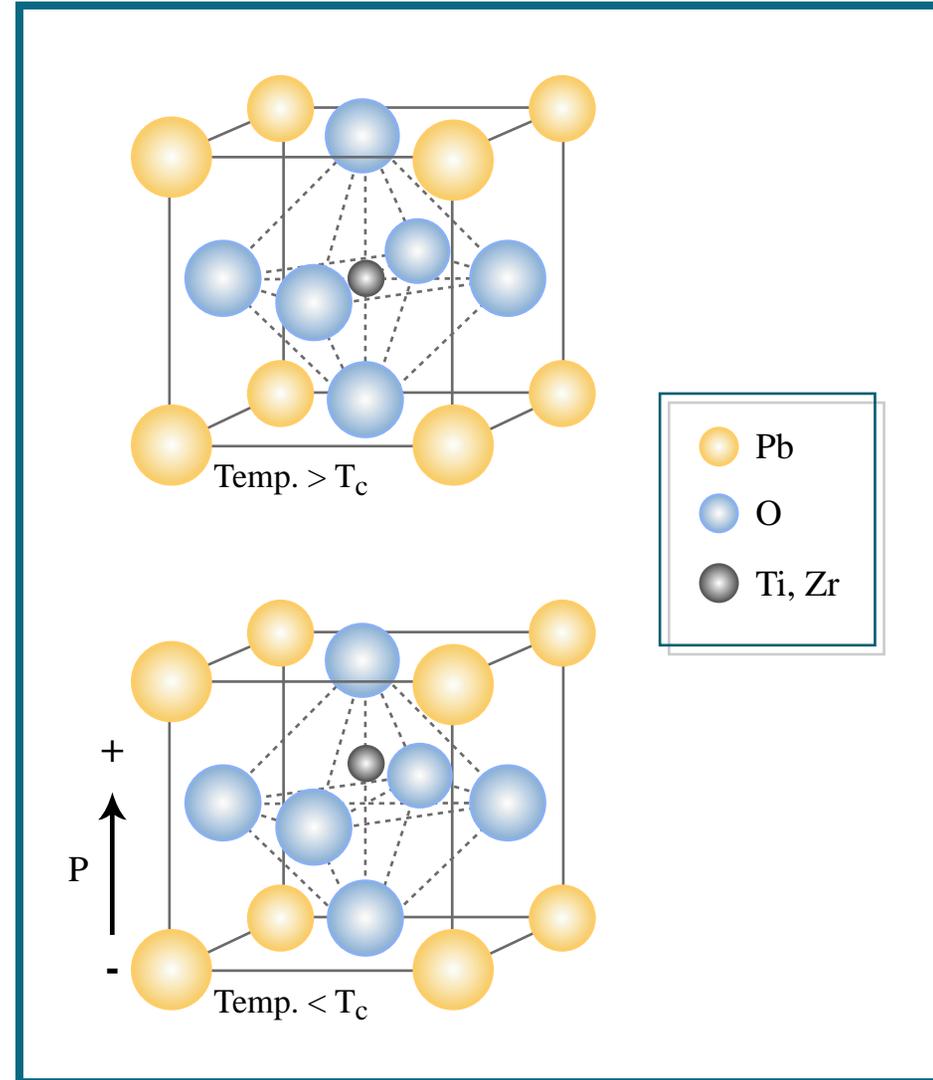


Figure by MIT OCW.

ex) $\text{Pb}(\text{Ti}_x\text{Zr}_{1-x})\text{O}_3$ (PZT)

Spontaneous polarization

$$\Delta \vec{P} = \vec{P}(\text{polar}) - \vec{P}(\text{nonpolar}) / a^2 c$$

: Polarization per unit cell volume

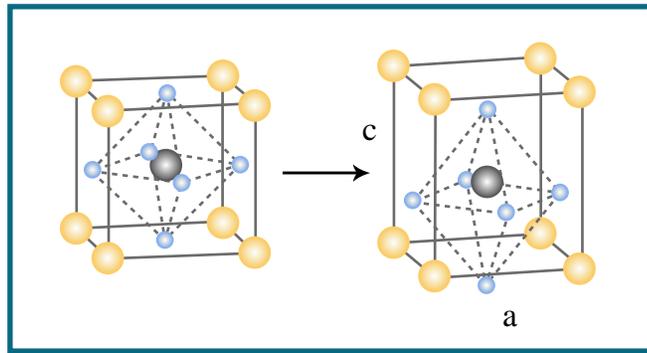
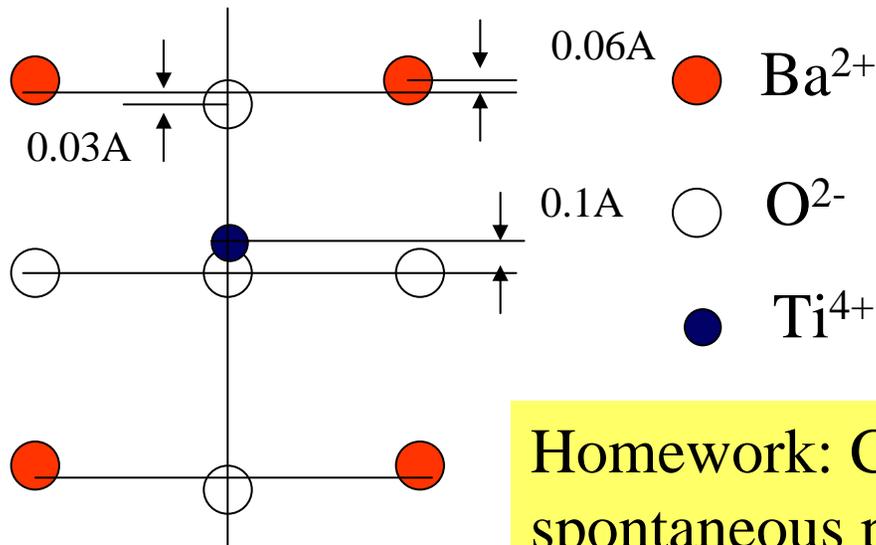
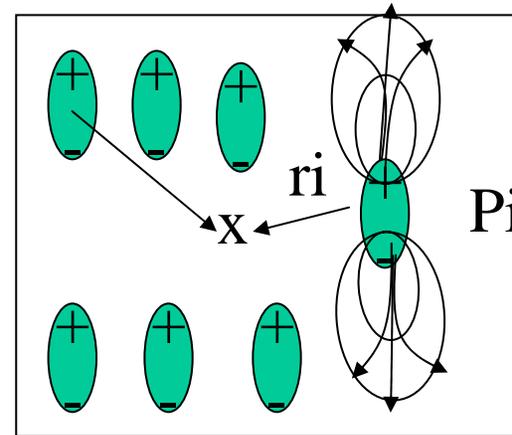


Figure by MIT OCW.



Homework: Calculate the magnitude of spontaneous polarization. $a=3.992 \text{ \AA}$, $c=4.036a$
 $1e=1.602 \times 10^{-19} \text{ C}$

Energy of polarization

